

PRACTICE 6 KEY – STOICH IN-CLASS REVIEW

My abbreviations...

TN = given or “teacher number”

CE = coefficients

MM = molar mass from periodic table

- 1) (TN) $\frac{\text{mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} = (\text{answer}) \text{ mol B}$
- 2) (TN) $\frac{\text{g A}}{(\text{MM}) \text{ g A}} \times \frac{1 \text{ mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} \times \frac{6.02 \times 10^{23} \text{ r.p. B}}{1 \text{ mol B}} = (\text{answer}) \text{ r.p. B}$
- 3) (TN) $\frac{\text{L A}}{22.4 \text{ L A}} \times \frac{1 \text{ mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{MM}) \text{ g B}}{1 \text{ mol B}} = (\text{answer}) \text{ g B}$
- 4) (TN) $\frac{\text{mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} \times \frac{6.02 \times 10^{23} \text{ r.p. B}}{1 \text{ mol B}} = (\text{answer}) \text{ r.p. B}$
- 5) (TN) $\frac{\text{r.p. A}}{6.02 \times 10^{23} \text{ r.p. A}} \times \frac{1 \text{ mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} \times \frac{22.4 \text{ L B}}{1 \text{ mol B}} = (\text{answer}) \text{ L B}$
- 6) (TN) $\frac{\text{L A}}{22.4 \text{ L A}} \times \frac{1 \text{ mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} = (\text{answer}) \text{ mol B}$
- 7) (TN) $\frac{\text{r.p. A}}{6.02 \times 10^{23} \text{ r.p. A}} \times \frac{1 \text{ mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{MM}) \text{ g B}}{1 \text{ mol B}} = (\text{answer}) \text{ g B}$
- 8) (TN) $\frac{\text{g A}}{(\text{MM}) \text{ g A}} \times \frac{1 \text{ mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} \times \frac{22.4 \text{ L B}}{1 \text{ mol B}} = (\text{answer}) \text{ L B}$
- 9) (TN) $\frac{\text{r.p. A}}{6.02 \times 10^{23} \text{ r.p. A}} \times \frac{1 \text{ mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} = (\text{answer}) \text{ mol B}$
- 10) (TN) $\frac{\text{L A}}{22.4 \text{ L A}} \times \frac{1 \text{ mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} \times \frac{6.02 \times 10^{23} \text{ r.p. B}}{1 \text{ mol B}} = (\text{answer}) \text{ r.p. B}$
- 11) (TN) $\frac{\text{mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{MM}) \text{ g B}}{1 \text{ mol B}} = (\text{answer}) \text{ g B}$
- 12) (TN) $\frac{\text{r.p. A}}{6.02 \times 10^{23} \text{ r.p. A}} \times \frac{1 \text{ mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} \times \frac{22.4 \text{ L B}}{1 \text{ mol B}} = (\text{answer}) \text{ L B}$
- 13) (TN) $\frac{\text{g A}}{(\text{MM}) \text{ g A}} \times \frac{1 \text{ mol A}}{(\text{CE}) \text{ mol A}} \times \frac{(\text{CE}) \text{ mol B}}{(\text{CE}) \text{ mol A}} = (\text{answer}) \text{ mol B}$